**DOCKET NO.:** UPFF-0004 / N2437 **Application No.:** 10/053,085

Notice of Non-Compliant Amendment Dated: June 16, 2009

PATENT REPLY FILED PURSUANT TO 37 CFR § 1.111

## **Listing of Claims**

This listing of claims supersedes all previous listings of claims.

## What is Claimed:

- 1. (Canceled)
- 2. (Currently amended) The fuel cell according to claim 62, wherein the hydrocarbon is a petroleum distillate, butane, toluene, decane, and mixtures thereof.
- 3. (Previously presented) The fuel cell according to claim 2, wherein the petroleum distillate is selected from the group consisting of gasoline, diesel oil, naphtha, JP-4, JP-5, JP-8, kerosene, motor oil, natural gas, fuel oil, and mixtures thereof.
- 4. (Previously presented) The fuel cell according to claim 3, wherein the petroleum distillate is selected from the group consisting of JP-4, JP-5, JP-8, and mixtures thereof.
- 5. (Previously presented) The fuel cell according to claim 3, wherein the petroleum distillate is selected from the group consisting of naptha, kerosene, fuel oil, and mixtures thereof.
- 6. (Previously presented) The fuel cell according to claim 3, wherein the petroleum distillate is selected from the group consisting of gasoline, diesel oil, natural gas, and mixtures thereof.
- 7. (Original) The fuel cell according to claim 2, wherein the hydrocarbon comprises an alcohol.
- 8. (Previously presented) The fuel cell according to claim 7, wherein the alcohol comprises ethanol.
- 9. (Previously presented) The fuel cell according to claim 2, wherein the hydrocarbon is

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selected from the group consisting of butane, toluene, decane, and mixtures thereof.

10. (Previously presented) The fuel cell according to claim 62, wherein the sulfur containing hydrocarbon fuel has a sulfur content of from about 1 ppm to about 1000 ppm.

11. (Previously presented) The fuel cell according to claim 10, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 10 ppm to about 1000 ppm.

12. (Previously presented) The fuel cell according to claim 11, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 20 ppm to about 1000 ppm.

13. (Previously presented) The fuel cell according to claim 12, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 100 ppm to about 1000 ppm.

14. (Previously presented) The fuel cell according to claim 13, wherein the sulfur-containing hydrocarbon fuel has a sulfur content of from about 250 ppm to about 1000 ppm.

15. (Previously presented) The fuel cell system according to claim 62 wherein the solid electrolyte is an oxide ion conducting material.

16. (Previously presented) The fuel cell according to claim 15, wherein the oxide ion conducting material is selected from the group consisting of doped ceria, doped zirconia, and doped lanthanum gallate.

17. (Previously presented) The fuel cell according to claim 16, wherein the doped ceria is selected from the group consisting of gadolinium doped ceria, samarium-doped ceria, yttria-doped ceria, and mixtures thereof.

18. (Previously presented) The fuel cell according to claim 15, wherein the oxide ion conducting material is yttria-doped zirconia.

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19. (Previously presented) The fuel cell according to claim 16, wherein the doped zirconia is

scandium-doped zirconia.

20. (Canceled)

21 (Previously presented) The process according to claim 63, wherein the hydrocarbon

is a petroleum distillate.

22. (Previously presented) The process according to claim 21, wherein the petroleum

distillate is selected from the group consisting of gasoline, diesel oil, naphtha, JP-4, JP-5,

JP-8, kerosene, motor oil, natural gas, fuel oil, and mixtures thereof.

23. (Previously presented) The process according to claim 22, wherein the petroleum

distillate is selected from the group consisting of JP-4, JP-5, JP-8, and mixtures thereof.

24. (Previously presented) The process according to claim 22, wherein the petroleum

distillate is selected from the group consisting of naphtha, kerosene, fuel oil, and mixtures

thereof.

25. (Original) The process according to claim 22, wherein the petroleum distillate comprises

gasoline.

26. (Original) The process according to claim 22, wherein the petroleum distillate comprises

diesel oil.

27. (Previously presented) The process according to claim 63, wherein the hydrocarbon

is selected from the group consisting of alcohols, butane, toluene, decane, and mixtures

thereof.

28. (Original) The process according to claim 27, wherein the hydrocarbon comprises an

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alcohol.

29. (Previously presented) The process according to claim 28, wherein the alcohol comprises ethanol.

30. (Previously presented) The process according to claim 63, wherein the sulfur containing hydrocarbon has a sulfur content of from about 10 ppm to about 1000 ppm.

31. - 53. (Canceled without prejudice)

54. (Canceled)

55. (Previously presented) The fuel cell system of claim 62, wherein the anode further comprises copper deposited in the pores.

56. (Previously presented) The process of claim 63, wherein the anode further comprises copper deposited in the pores.

57. (Canceled)

58. (Canceled) The fuel cell system of claim 64, wherein the anode further comprises copper deposited in the pores.

59. (Canceled)

60. (Canceled) The process of claim 66, wherein the anode further comprises copper deposited in the pores.

61. (Canceled)

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62. (Previously presented) A solid oxide fuel cell system capable of directly operating with a sulfur-containing hydrocarbon fuel that does not undergo prior treatment to remove organic

sulfur compounds, comprising:

(a) a solid electrolyte comprising an electronic insulator that allows transfer of

anions;

(b) an essentially nickel-free porous anode containing at least ceria deposited in the

pores, the anode further comprising a ceramic, and at least a portion of the anode being bound

to the electrolyte;

(c) a cathode;

(d) a fuel comprising a hydrocarbon having 2 or more carbons, and the fuel being

characterized as having a sulfur content of from about 1 ppm to about 5000 ppm; and

(e) an oxygen source;

wherein the solid electrolyte and the porous anode overlap one another so as to define

a region of physical contact between one another, the region of physical contact being

characterized as an essentially uninterrupted interface.

63. (Previously presented) A process of producing electrical energy, comprising:

(a) providing a solid oxide fuel cell system capable of directly operating with a sulfur-

containing hydrocarbon fuel that does not undergo prior treatment to remove organic

sulfur compounds, the solid oxide fuel cell system comprising:

a solid oxide electrolyte comprising an electronic insulator that allows transfer

of anions;

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an essentially nickel-free porous anode containing at least ceria deposited in

the pores, the anode further comprising a ceramic, and at least a portion of the

anode being bound to the electrolyte; and

a cathode, and

a fuel comprising a hydrocarbon having two or more carbons, and the fuel

being characterized as having a sulfur content of from about 1 ppm to about

5000 ppm,

wherein the solid electrolyte and the anode overlap one another so as to define a

region of physical contact between one another, the region of physical contact being

characterized as an essentially uninterrupted interface,

(b) contacting the cathode with an oxygen source; and

(c) contacting the porous anode with the fuel.

64. (Previously presented) A solid oxide fuel cell system capable of directly operating with a

sulfur-containing hydrocarbon fuel that does not undergo prior treatment to remove organic

sulfur compounds, comprising:

(a) a solid electrolyte comprising an electronic insulator that allows transfer of

anions;

(b) an essentially nickel-free porous anode containing at least copper deposited in the

pores, the anode further comprising a ceramic, and at least a portion of the anode being bound

to the electrolyte;

(c) a cathode;

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(d) a fuel comprising a hydrocarbon having 2 or more carbons, and the fuel being

characterized as having a sulfur content of from about 1 ppm to about 5000 ppm; and

(e) an oxygen source;

wherein the solid electrolyte and the porous anode overlap one another so as to define

a region of physical contact between one another, the region of physical contact being

characterized as an essentially uninterrupted interface.

65. (Canceled)

66. (Canceled) A process of producing electrical energy, comprising:

(a) providing a solid oxide fuel cell system capable of directly operating with a sulfur-

containing hydrocarbon fuel, the solid oxide fuel cell comprising

a solid oxide electrolyte that is an electronic insulator that allows transfer of anions,

an essentially nickel-free porous anode, the anode further comprising a ceramic, the

anode containing at least copper ceria deposited in the pores and comprising a porous

ceramic, and at least a portion of the anode being bound to the electrolyte, and

a cathode,

(b) contacting said cathode with an oxygen source; and

(c) contacting said porous anode with a fuel comprising a hydrocarbon having two or more

carbons, the fuel being characterized as having a sulfur content of from about 1 ppm to about

5000 ppm,

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wherein the solid electrolyte and the porous anode overlap one another so as to define a region of physical contact between one another, the region of physical contact being characterized as an essentially uninterrupted interface.

67. (Canceled)

68. (Canceled) The process according to claim 66, wherein the hydrocarbon is selected from the group consisting of alcohols, a petroleum distillate, butane, toluene, decane, or any combination thereof.